



Contribution ID : 38

Type : Poster

## The Relationship of Li<sup>+</sup> Displacement and Temperature in the Garnet Oxides Li<sub>7-x</sub>La<sub>3</sub>Zr<sub>2-x</sub>Ta<sub>x</sub>O<sub>12</sub>

The lithium containing garnet oxides of Li<sub>7</sub>La<sub>3</sub>M<sub>2</sub>O<sub>12</sub> (M = Ta, Nb) were reported to have Li<sup>+</sup> conducting abilities back in 2003.<sup>1</sup> Since then lithium containing garnet oxides have been identified as a electrolyte candidate for lithium ion batteries as they exhibit the physical and chemical properties desired for a solid-state electrolytes.<sup>2</sup> The most notable garnet oxide material is the cubic phase (Ia-3d) Li<sub>6.4</sub>La<sub>3</sub>Zr<sub>1.4</sub>Ta<sub>0.6</sub>O<sub>12</sub> which has a reported Li<sup>+</sup> conductivity of  $1 \times 10^{-3}$  S cm.<sup>1,3</sup> Here the structures and Li<sup>+</sup> site occupancies of garnet series of Li<sub>7-x</sub>La<sub>3</sub>Zr<sub>2-x</sub>Ta<sub>x</sub>O<sub>12</sub> (x = 0, 0.25, 0.50, 0.75, 1.00) was further explored. A one step heating synthesis was employed for garnet preparing. Successful cubic phase garnet synthesis was confirmed with X-ray powder diffraction. Determination of lithium occupancy at the Li<sub>24</sub> and Li<sub>96</sub> sites was achieved with high-resolution neutron powder diffraction. NPD data were collected at room temperature, 200°C, 400°C and 600°C.

### References

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### Topic

Advanced Materials

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**Session Classification** : Poster Session

**Track Classification** : Advanced Materials