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Nucleon-based science benefiting all Australians

NEW USER SYMPOSIUM 2016
6 SEPTEMBER 2016
National Centre for Spectroscopy Science

Neutron Small Angle Scattering

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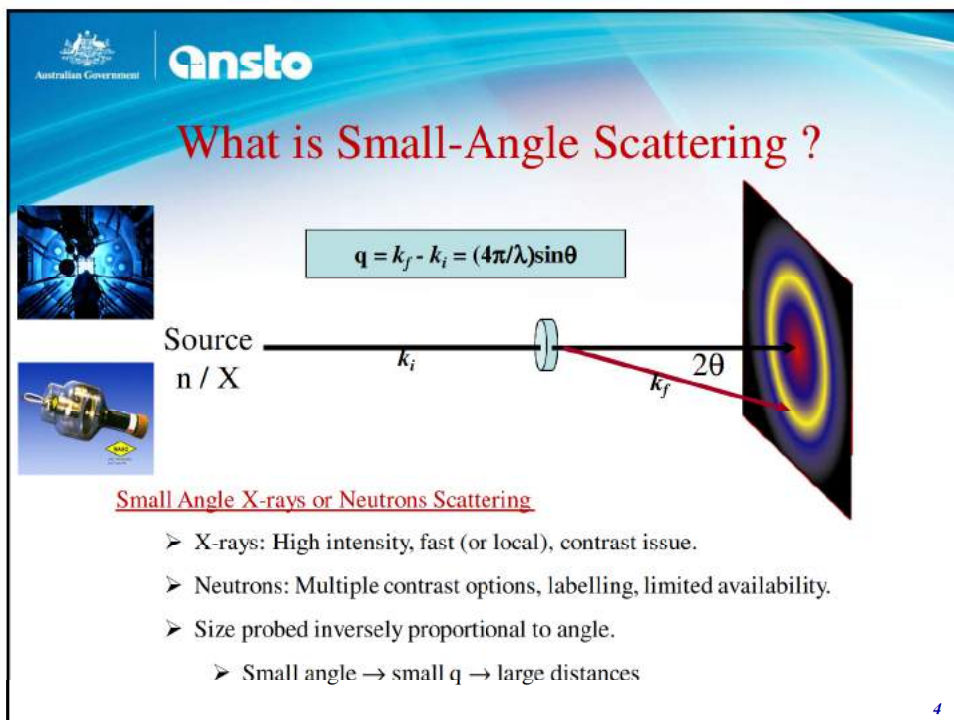
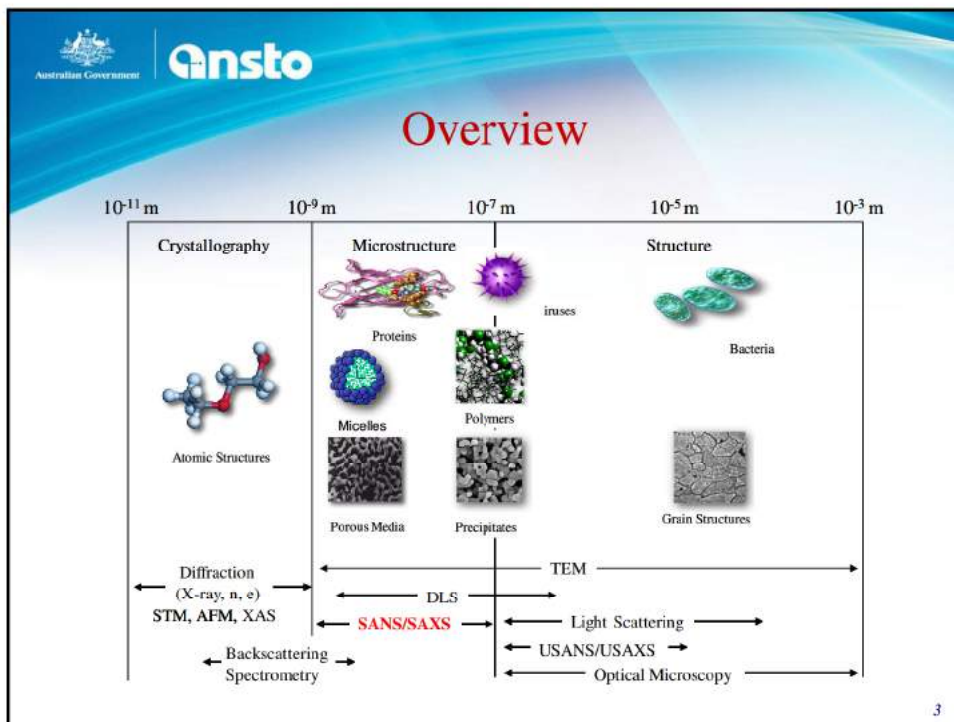
6th September 2016

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Outline

- Introductions: What is SAS? Why Neutrons?
- Instruments: SANS (Quokka, Bilby-TOF), SAXS, and USANS
- Quokka conventional pin-hole SANS
- Science with SANS
- Sample environments
- Recent examples
- Take home message

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Why Neutrons?

- Neutrons have zero charge and negligible electric dipole → interact with matter via nuclear forces.
- Nuclear forces are very short range.
- Highly penetrating.
- Attenuation of low energy neutrons by Al is ~1%/mm compared to >99%/mm for x-rays.
- Different interaction – “sees” different things
- Non-destructive


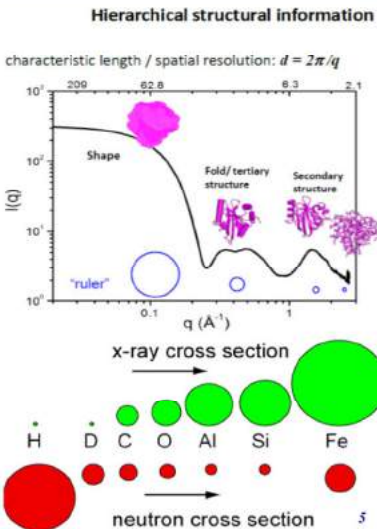


Diagram showing a neutron (left) and a tennis ball (right) for size comparison.



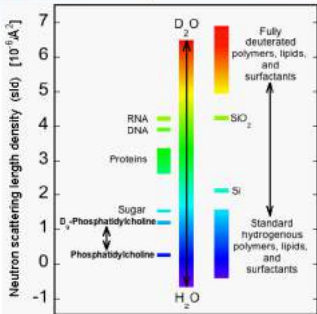
Hierarchical structural information
characteristic length / spatial resolution: $d = 2\pi/q$

The graph plots intensity $I(q)$ on a log scale from 10^1 to 10^3 against scattering vector q in \AA^{-1} from 0.1 to 2.1. Key features include a 'ruler' at $q \approx 0.1$, 'Shape' at $q \approx 0.2$, 'Fold/tertiary structure' at $q \approx 0.5$, and 'Secondary structure' at $q \approx 1.0$. Below the graph, a diagram shows 'x-ray cross section' (green circles) and 'neutron cross section' (red circles) for various elements: H, D, C, O, Al, Si, Fe.

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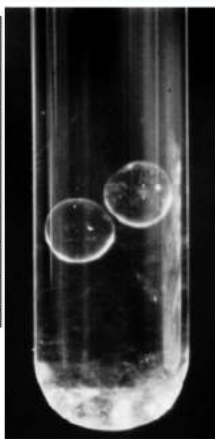
Contrast Variation

- Neutron scattering: Contrast Variation
 - protein – lipid
 - Solute – surfactant
 - protein – sugar
 - solute-solvent
 - part of molecules



Neutron scattering length density (sld) [10^6\AA^{-3}]

Scale from -1 to 7. Key points: H_2O at -1, Phosphatidylcholine at 0, Sugar at 1, D_2O at 2, RNA/DNA at 3, Proteins at 4, SiO_2 at 5, Fully deuterated polymers/lipids/surfactants at 6-7.



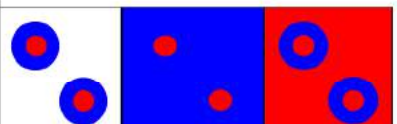


Diagram illustrating contrast variation in a mixture of particles (red and blue) in a solvent (blue and red).

G.D. Wignall, Physical properties of polymers, 1993, 424-502. 6

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SAS Instruments @ ACNS

Quokka

Bilby

Kookaburra

SANS

SAXS

USANS

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Quokka

- Quokka* is a pin-hole SANS instrument
- Located on the cold guide (CG1) at OPAL
- Q range from $6 \times 10^{-4} \text{ \AA}^{-1}$ to greater than 0.7 \AA^{-1}
- MgF_2 -Focussing optics
- Estimated maximum flux on sample is $> 2 \times 10^7 \text{ n/cm}^2/\text{s}$
- Astrium velocity selectors: 7 - 18% resolution
- Pre-sample collimation system: neutron guides and apertures
- Ordela 2D position sensitive detector (1 m² area and 5 mm each pixel resolution)

neutrons

x-translation
OR
autochanger

variable length source
terminated by
sample aperture

rotation

z-stage

position encoded rail

40 m

Cold Source

Detector
1x1m²
5mm/pixel

Velocity selector 11% - 18%

$$Q = \frac{4\pi \sin\theta}{\lambda}$$

* E.P. Gilbert, J.C. Schulz and Terry J. Noakes, "Quokka" - the Small-Angle Neutron Scattering Instrument at OPAL., Physica B, 385-386 (2006) 1180-1182.

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Science with SANS

- Becomes an all-purpose structural tool
- Characterizes a variety of samples:
 - Polymers, food, proteins, metals, ceramics, rocks, magnetic clusters, superconductors etc.
 - Part of molecule/system, multicomponent systems, aggregation etc.
- Complementary to electron microscopy
- SANS can provide particle sizes, shapes, and interaction
- Studies molecules under most relevant conditions
- Requires relative smaller amount of sample
- Fast data acquisition and analysis

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The Basic Scattering Equation

For an ensemble of identical, randomly oriented particles, the intensity of coherently, elastically scattered radiation is dependant only upon the magnitude of q , and can be expressed as:

$$I(q) = N(\Delta\bar{\rho}V)^2 P(q)S(q)$$

N = molecules/unit volume
 V = molecular volume
 $\Delta\bar{\rho} = \langle \rho(\mathbf{r}) \rangle - \bar{\rho}_s$ = contrast, the scattering density difference between the scattering particle and solvent
 $P(q)$ = form factor \Rightarrow particle shape
 $S(q)$ = structure factor \Rightarrow inter-particle correlation distances

characteristic length / spatial resolution: $d = 2\pi/q$

http://qr5.org/index_of_refraction

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Sample Environments

- Two automatic temperature-controlled sample changers: -40 to 120 °C or between room temperature and 250 °C
- New 12 position individual temperature controlled sample changer
- Superconducting magnet up to 11 T and temperatures of <1 K
- A BioLogic SFM-300 stopped flow cell
- Rheo-SANS setup
- 5 position rotating sample changer
- Various sample holders for different applications
- A Rapid Heat Quench Cell: For thermal shock studies
- The neutron Rapid Visco Analyser (nRVA)
- High pressure SANS cell
- In-situ Differential Scanning Calorimetry (DSC)

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Recent Examples

Effects of Crowding and Environment on the Intrinsically Disordered Protein, Rec1-Resilin*

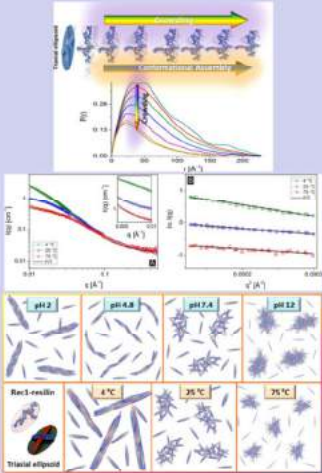
The structural ensembles and transitions of a biomimetic, multi-stimuli-responsive, intrinsically disordered protein (IDP), Rec1-resilin with change in molecular crowding and environment (concentration, pH, and temperature) is evaluated using SANS and SAXS.

Rec1-resilin in uncrowded solutions → radius of gyration $R_g \sim 5$ nm, and a scattering function for the triaxial ellipsoidal model best fit model.

On crowding (increase in concentration >10 wt %), Rec1-resilin molecules exert intermolecular repulsive force of interaction, the R_g value reduces with a progressive increase in concentration.

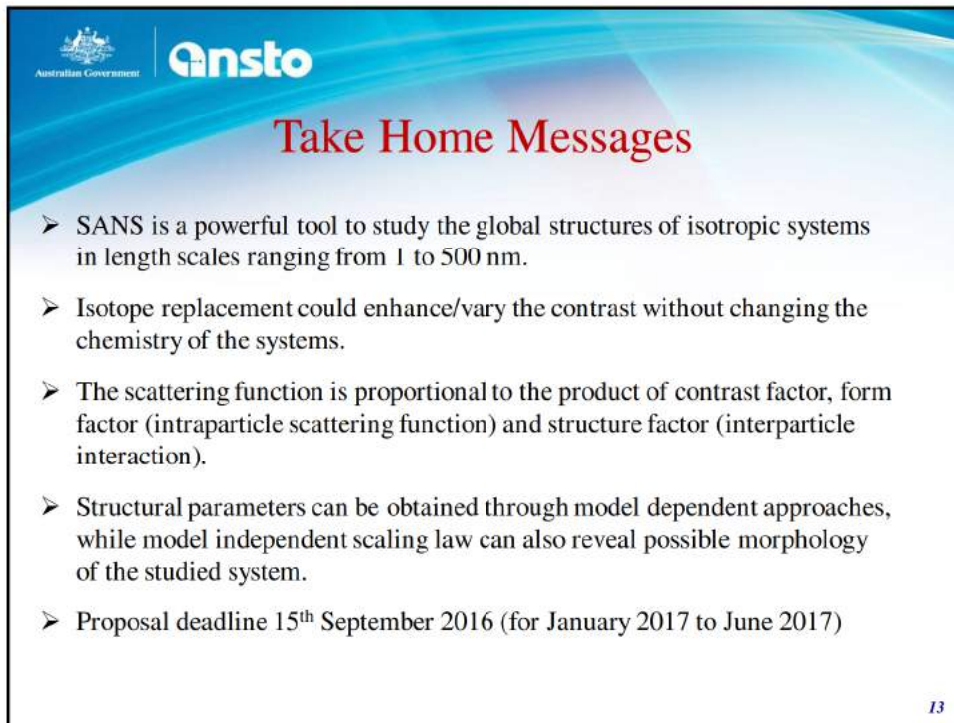
The structural organization of Rec1-resilin dynamically transforms from a rod (pH 2) to coil (pH 4.8) and to globular (pH 12) as a function of pH.

The temperature-triggered dual-phase-transition: rod-shaped below the upper critical solution temperature (~ 4 °C) and a large but compact structure above the lower critical solution temperature (~ 75 °C).



*Balu, R., Meta, J. P., Knott, R., Elvin, C. M., Hill, A. J., Choudhury, N. R., Dutta, N. K. (2016) *J. Phys. Chem. B* **120**, 6490–6503.

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Take Home Messages

- SANS is a powerful tool to study the global structures of isotropic systems in length scales ranging from 1 to 500 nm.
- Isotope replacement could enhance/vary the contrast without changing the chemistry of the systems.
- The scattering function is proportional to the product of contrast factor, form factor (intraparticle scattering function) and structure factor (interparticle interaction).
- Structural parameters can be obtained through model dependent approaches, while model independent scaling law can also reveal possible morphology of the studied system.
- Proposal deadline 15th September 2016 (for January 2017 to June 2017)

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Questions?

