

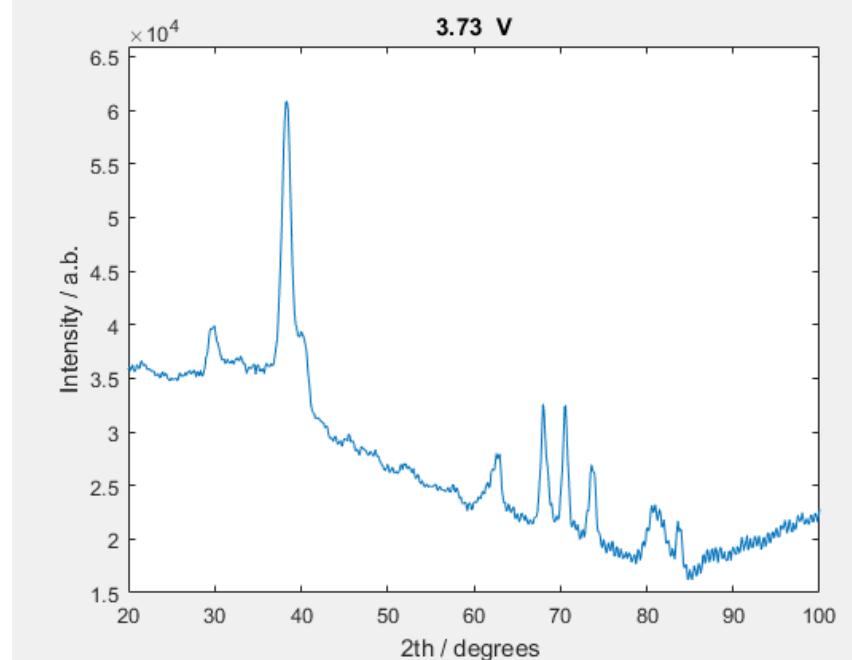
Developing better energy storage materials and devices with XAS

Never Stand Still

Science

Neeraj Sharma

neeraj.sharma@unsw.edu.au





Outline

- An overview of my research field
 - Battery materials research and development
- Examples of how we have used XAS
- What we have learnt
- Cutting edge experiments (*operando* and *in situ* XAS)
 - Thermal expansion materials (ceramics)
- Examples of how we are using XAS
- Multiscale information – critical for future development

BASIC HUMAN NEEDS

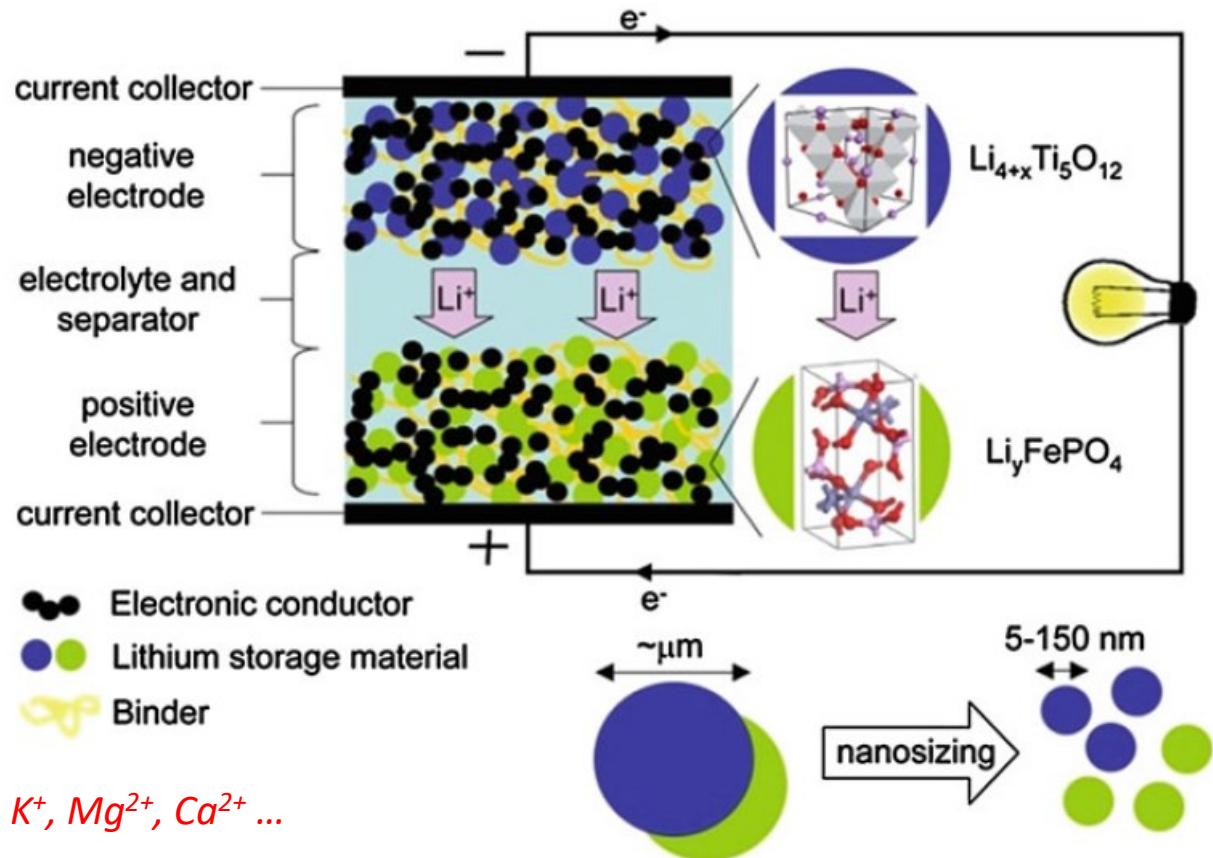
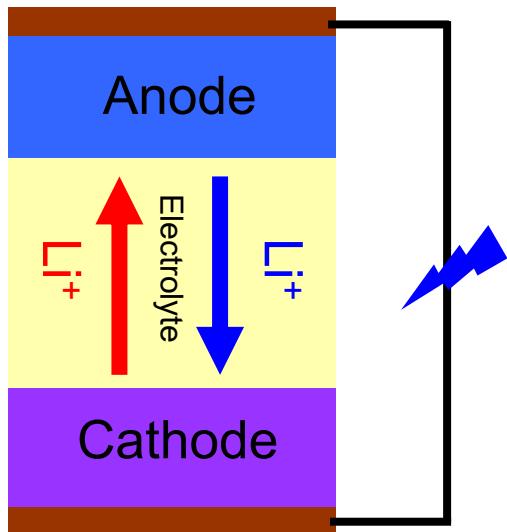


The Nobel Prize in Chemistry 2019 was awarded jointly to John B. Goodenough, M. Stanley Whittingham and Akira Yoshino "for the development of lithium-ion batteries."



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What does a lithium-ion battery really look like inside?



One can replace Li^+ with Na^+ , K^+ , Mg^{2+} , Ca^{2+} ...

Changes that occur (pertinent to XAS) – crystal structure, oxidation state, local environment

XAS is an awesome tool to use!



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Crystal structure

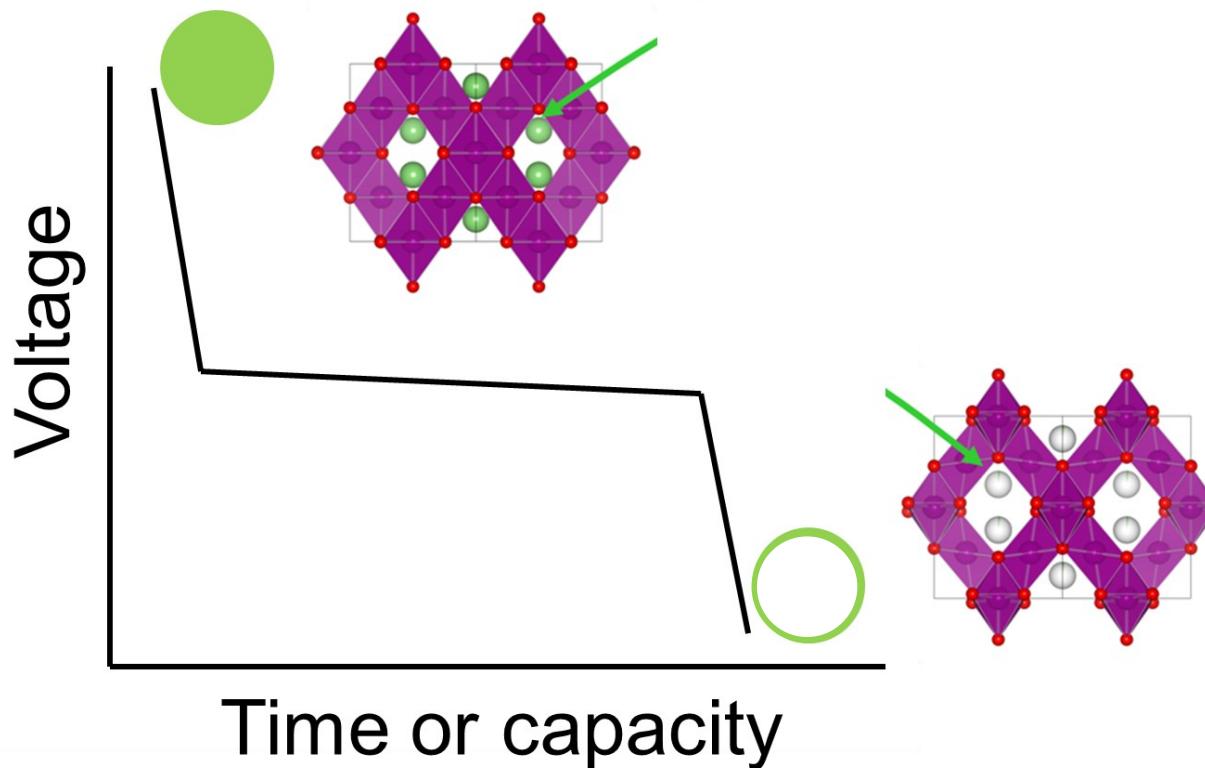
Evolution

Oxidation state

Local structure

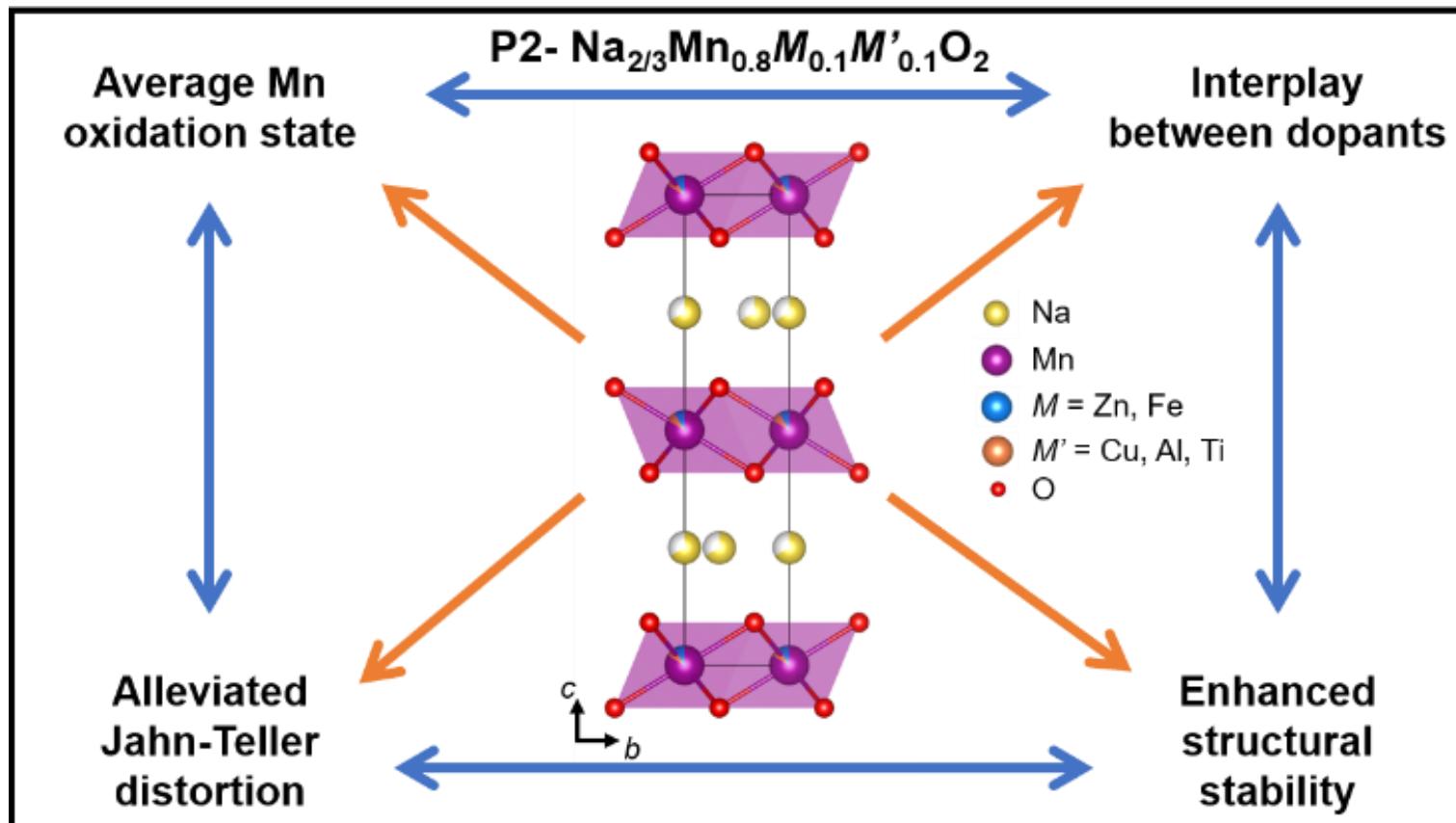
Reaction mechanisms

Properties



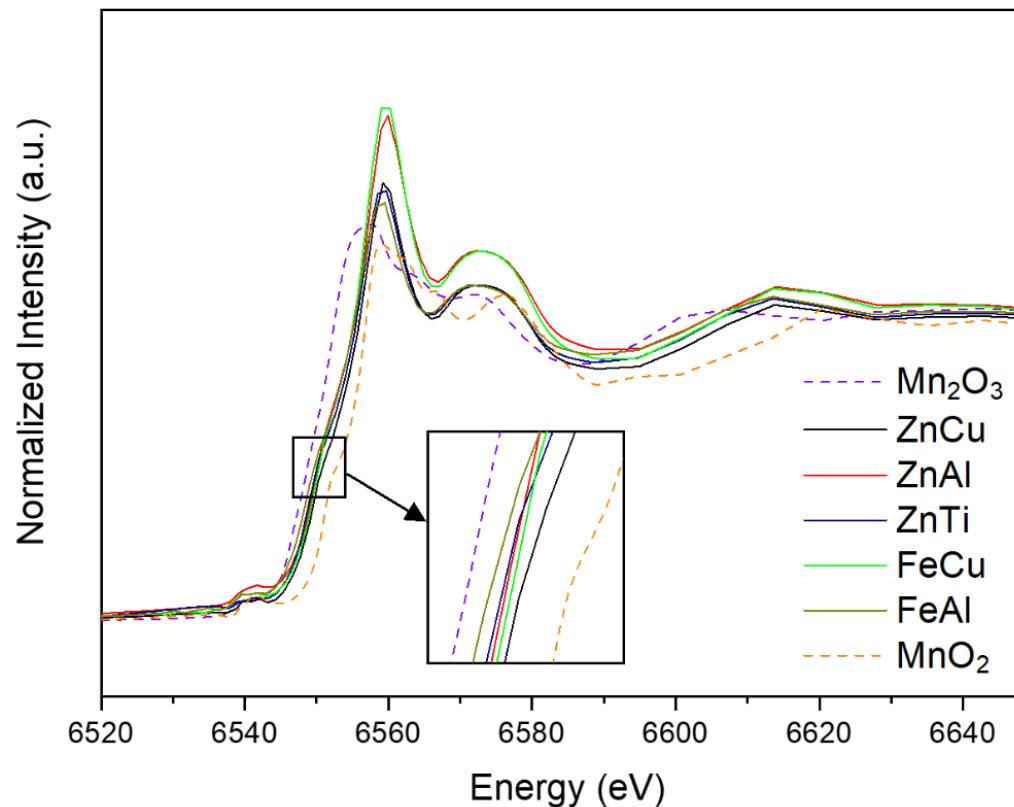
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An example: $\text{Na}_x\text{Mn}_{0.8}\text{TM}_{0.2}\text{O}_2$

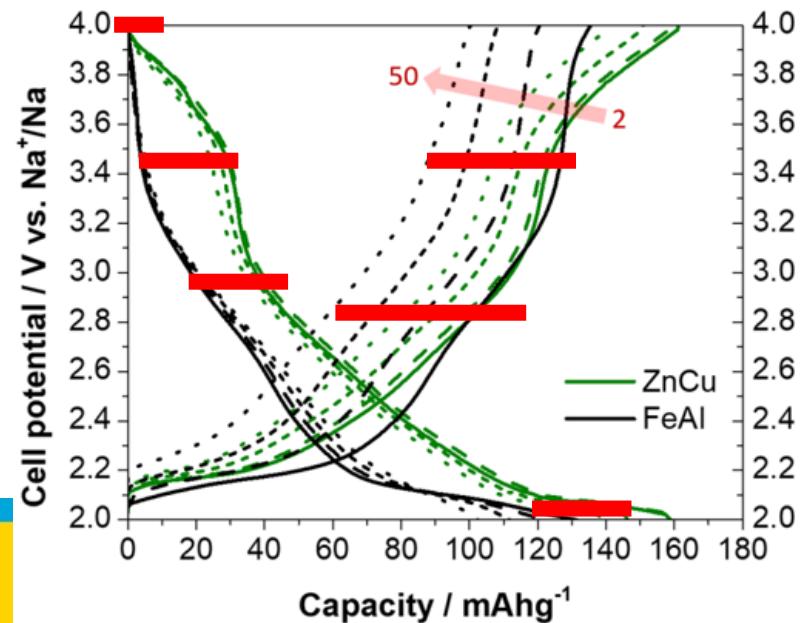
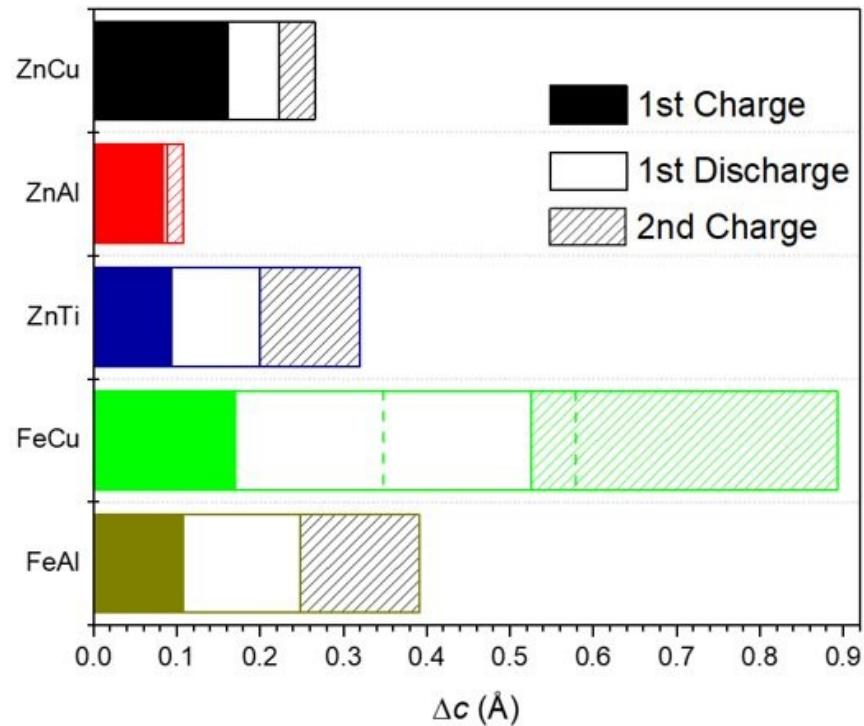
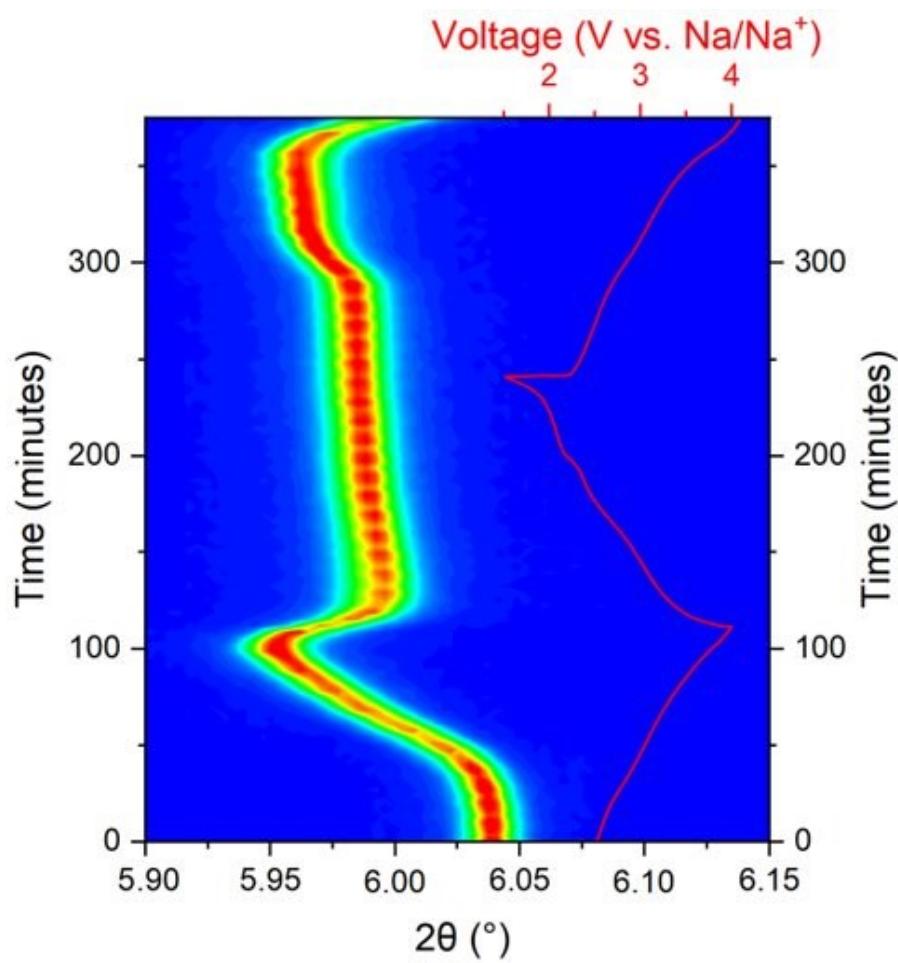


XAS to confirm Mn oxidation state trend

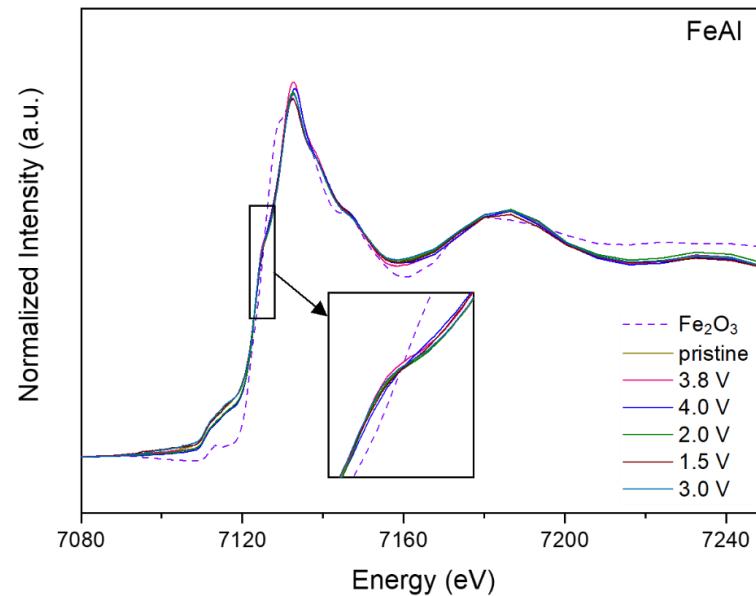
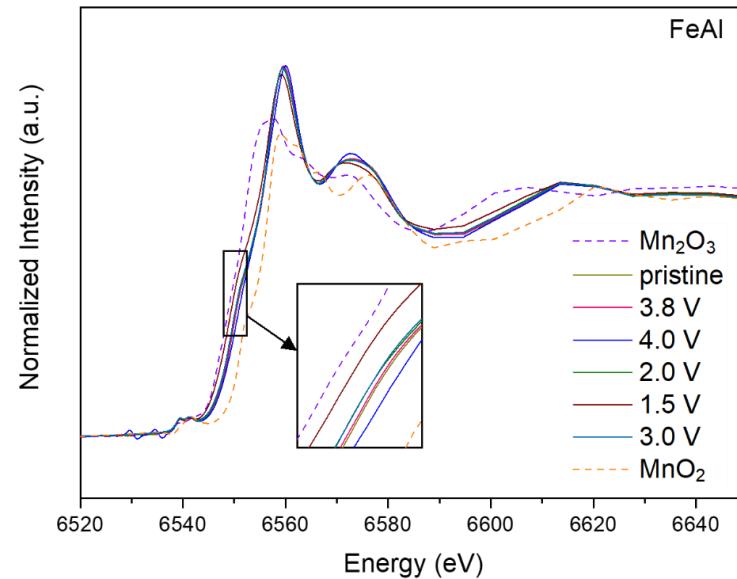
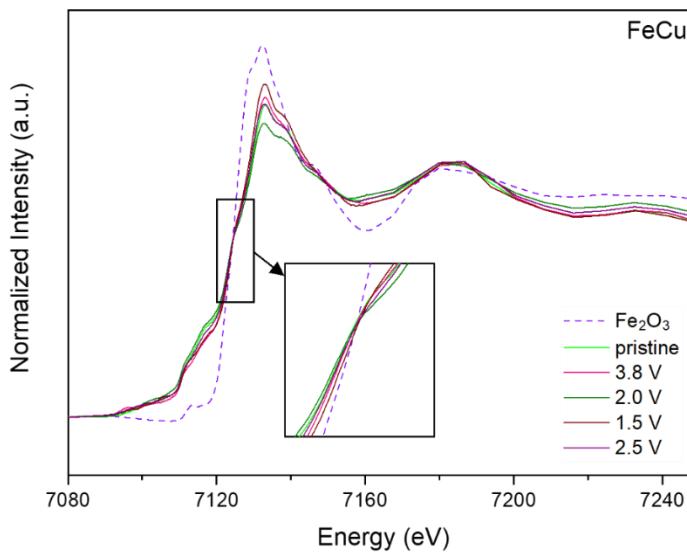
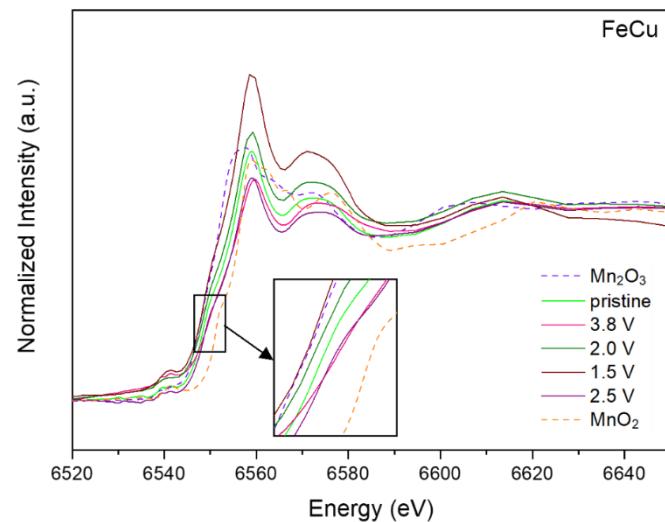
Chemical formula	Abbre-viation	Theoretical Mn oxidation state
$\text{Na}_{2/3}\text{Mn}_{0.8}\text{Zn}_{0.1}\text{Cu}_{0.1}\text{O}_2$	ZnCu	3.66
$\text{Na}_{2/3}\text{Mn}_{0.8}\text{Zn}_{0.1}\text{Al}_{0.1}\text{O}_2$	ZnAl	3.54
$\text{Na}_{2/3}\text{Mn}_{0.8}\text{Zn}_{0.1}\text{Ti}_{0.1}\text{O}_2$	ZnTi	3.41
$\text{Na}_{2/3}\text{Mn}_{0.8}\text{Fe}_{0.1}\text{Cu}_{0.1}\text{O}_2$	FeCu	3.54
$\text{Na}_{2/3}\text{Mn}_{0.8}\text{Fe}_{0.1}\text{Al}_{0.1}\text{O}_2$	FeAl	3.41



Structural Evolution



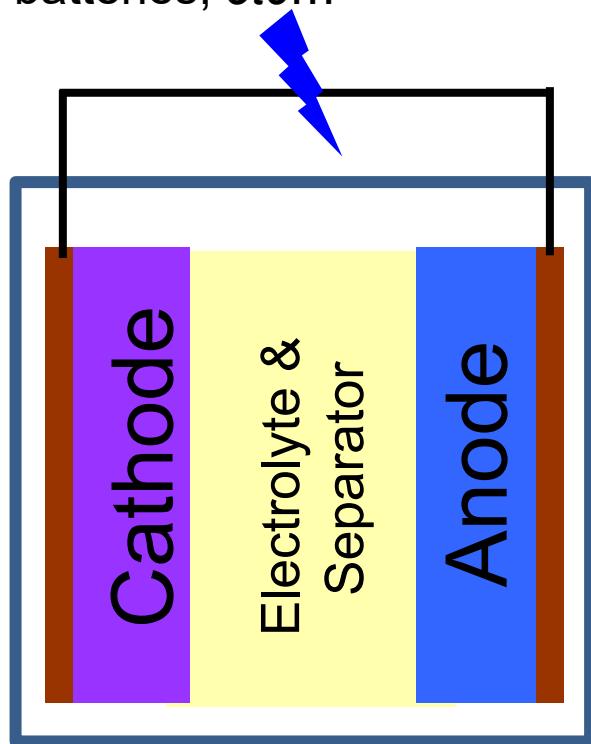
Ex situ XAS



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“In situ experiments” in devices....

Fuel cells, batteries, etc...



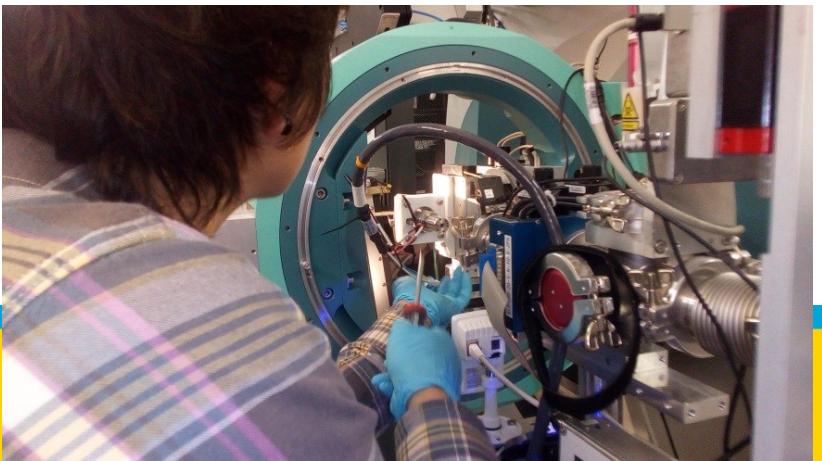
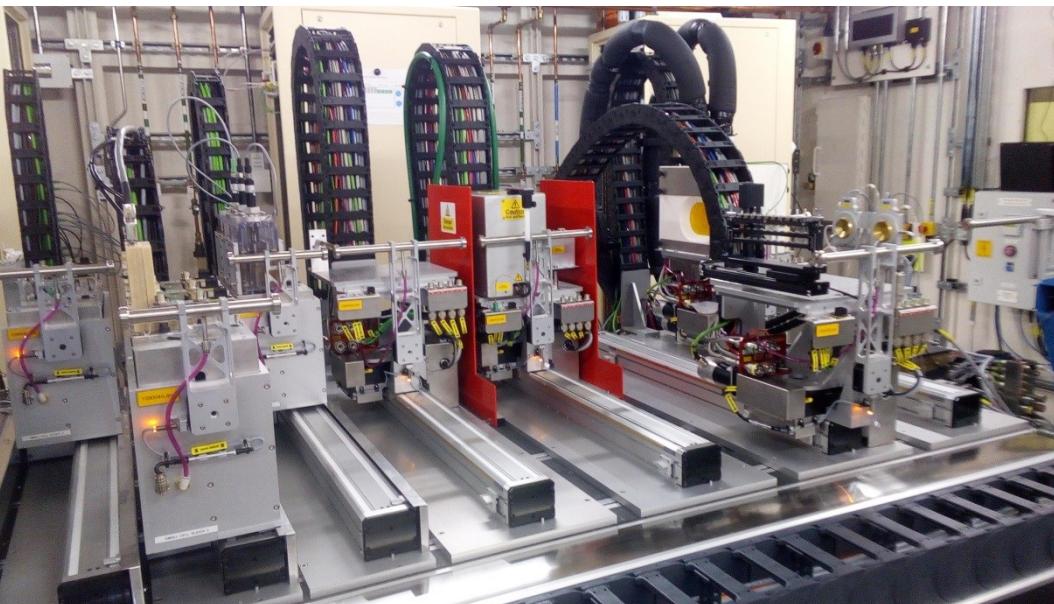
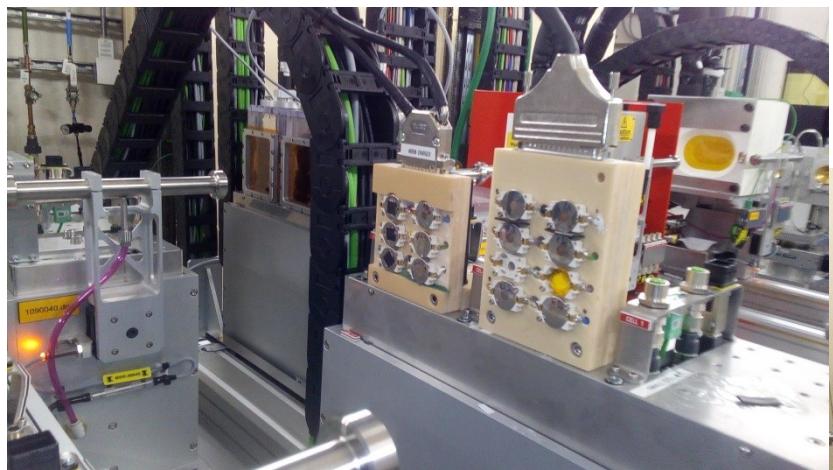
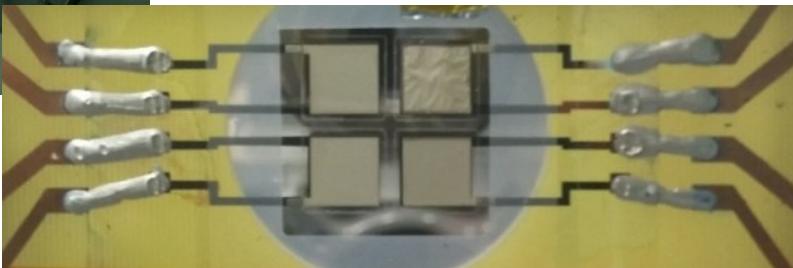
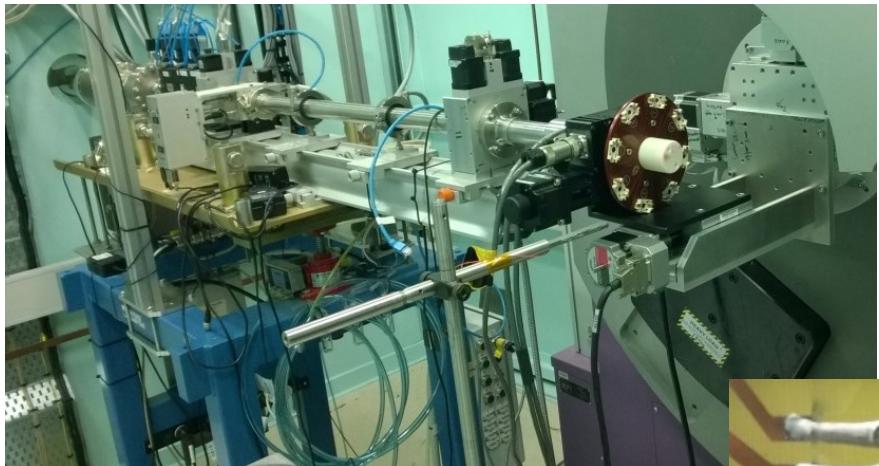
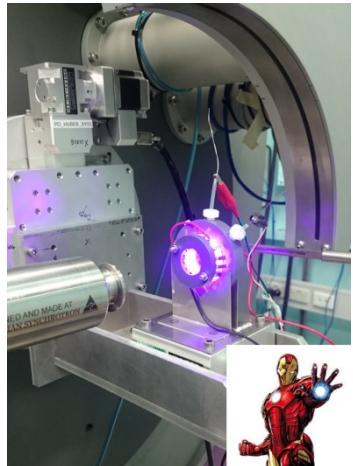
- Structure as a function of device performance (e.g. electrochemistry)
- Device performance at different temperatures (e.g. electrochemistry at LT, RT, HT)

Electrode mix = **active material**, carbon,
PVDF/PFTE

Current collectors (Al, Cu), stainless steel casing

Electrolyte = salt in carbonate solution

Separator = glass fibre, polyethylene

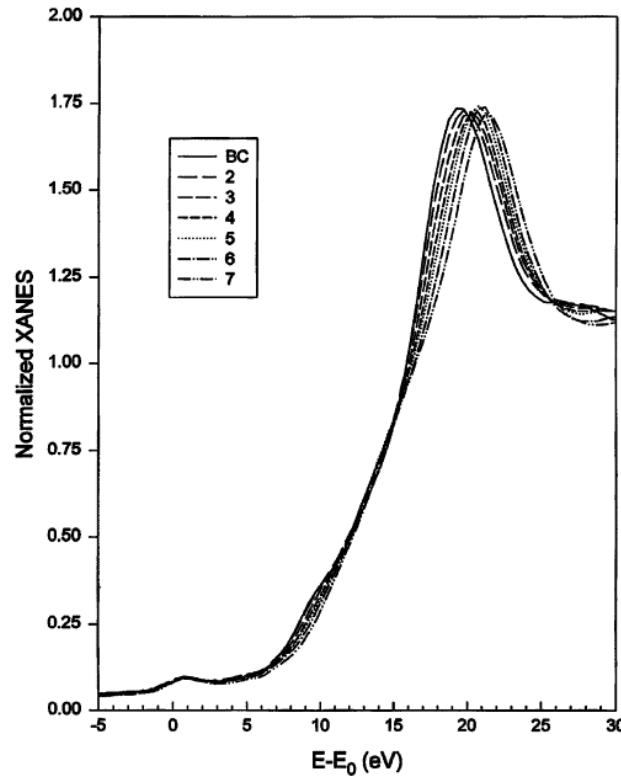
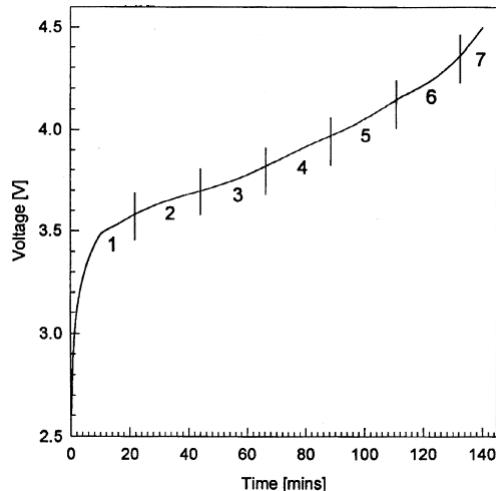
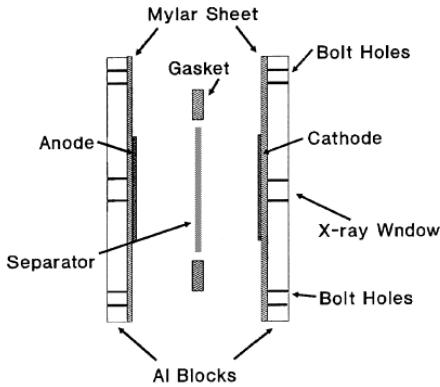


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It would be so cool to do *in situ* XAS

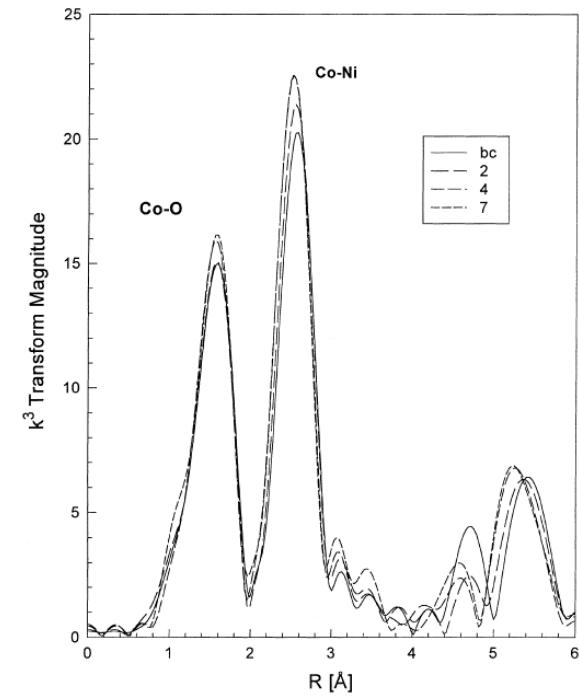
Current Australian Synchrotron beamlines – not so good at this*
(beam-based phenomena)

- International synchrotrons
- MEX



Co K edge

Journal of Power Sources 92 (2001) p1-8



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Some recent work – so called conversion electrode **BiSb**

- Very difficult to probe with diffraction
- Upon (de)lithiation nanomaterials are formed

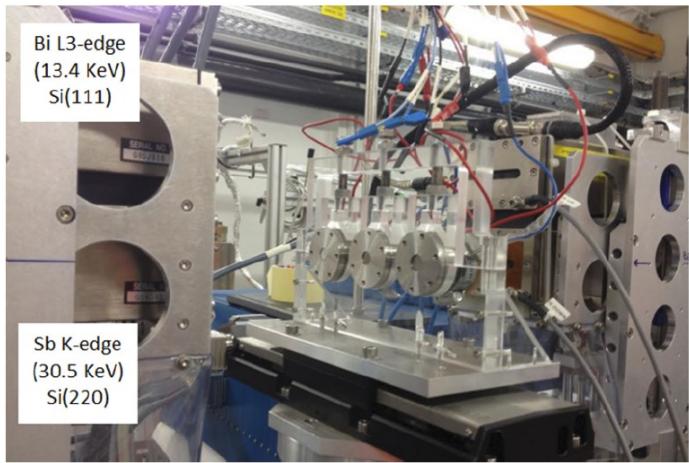
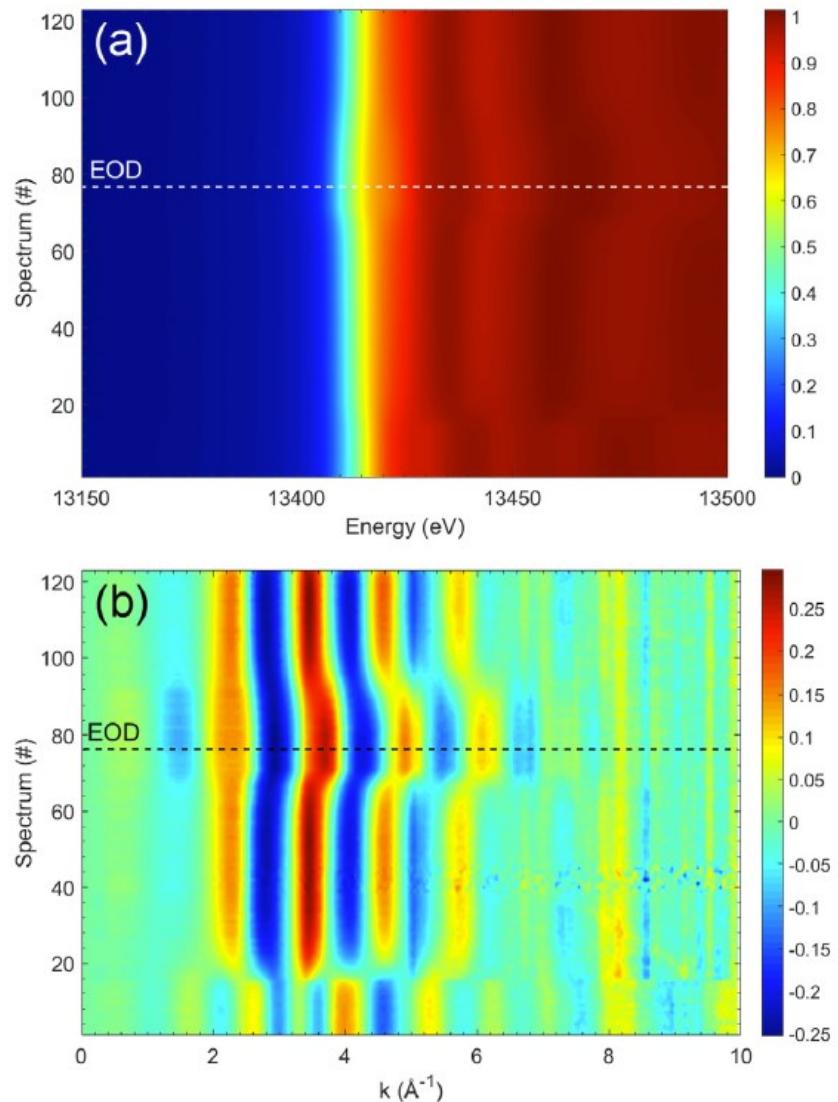


Figure 1: Experimental sample stage used for the simultaneous study of three electrochemical cells by *operando* Sb K-edge and Bi L₃-edge XAS.

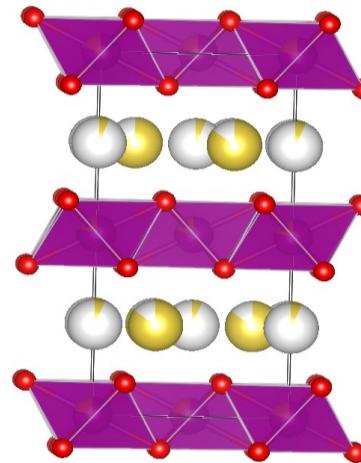
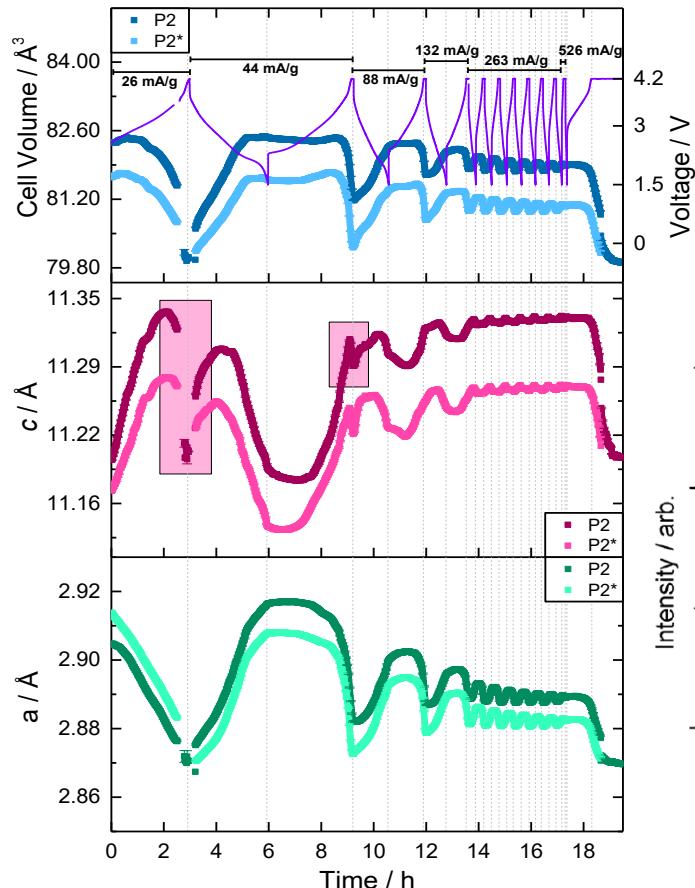
Principal component analysis



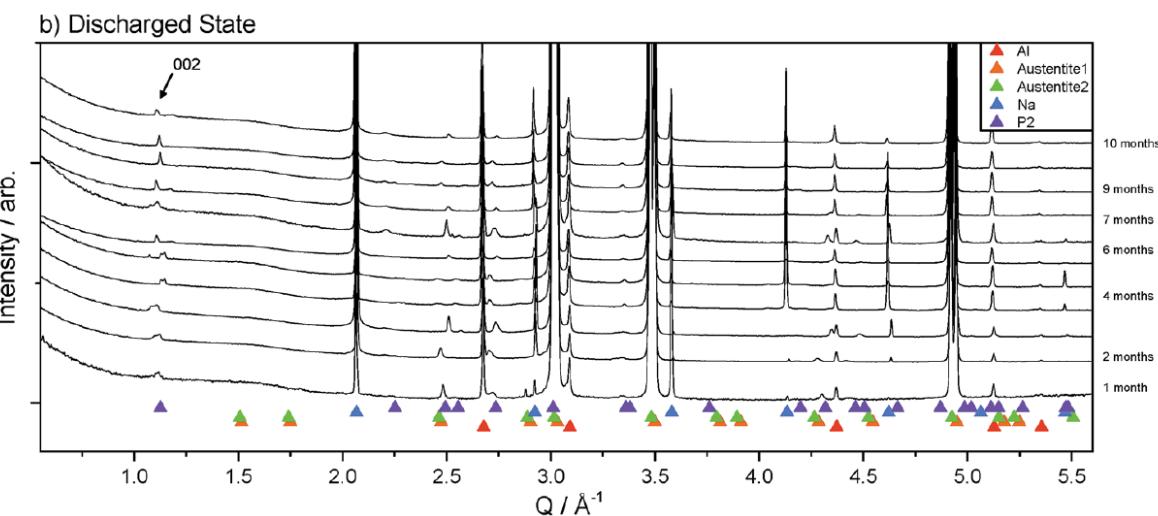
Long term and fast cycling experiments

P2- $\text{Na}_{0.67}\text{Mn}_{0.8}\text{Fe}_{0.1}\text{Ti}_{0.1}\text{O}_2$

Fast experiments 45 s/dataset
@ALBA



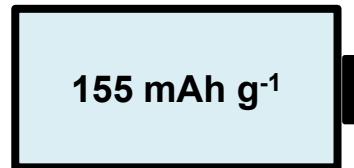
Long duration ~ 2 years @ Diamond



Looking at beyond lithium-ion batteries

E.g. Li-S batteries

Conventional Li-ion

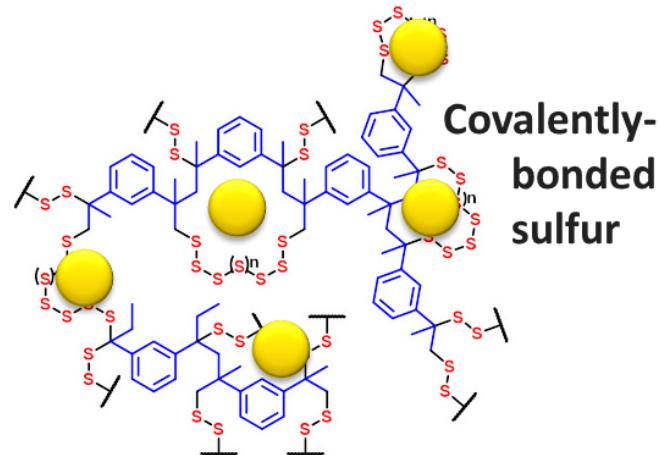


Li-S

1672 mAh g⁻¹
≈ 10x greater



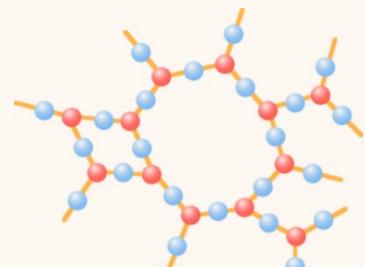
Sulfur-rich copolymers



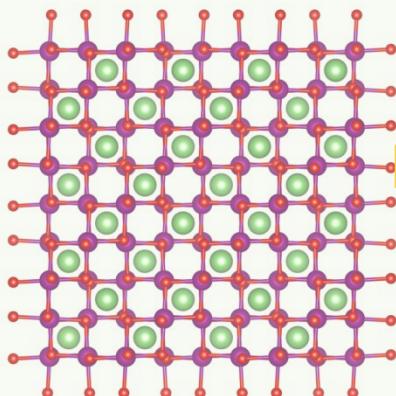
Scale is critical...

Length

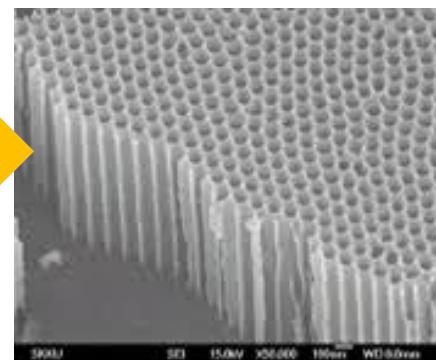
Local information



Long-range order



Nanoscale ordering



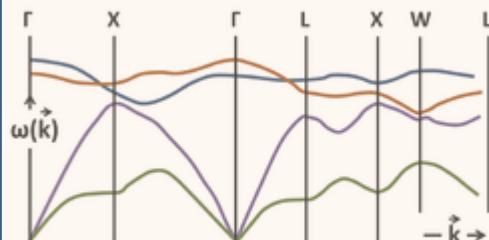
Fabrication



Surface considerations

Dynamics

Lattice vibrations



Ion hopping and concerted motion

Grotthuss mechanism (Proton hopping)



Vehicle mechanism



Bulk conductivity
(grain boundaries)

$$\sigma_{\text{e}} - \sigma_{\text{ion}}$$



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Uttam Mittal

PhD

Michael Fenech

Lisa Djuandhi

Matthew Teusner

Honours

Liam McKinlay

Rajko Romic

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PhD

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